1α,25-Dihydroxyvitamin D3

Active form of vitamin D

 $1\alpha,25$ -Dihydroxyvitamin D $_3$ is the active hormonal form of vitamin D. It inhibits proliferation and induces differentiation of a variety of cells including osteoclasts and HL-60 cells. It is a central regulator of bone and calcium homeostasis. It inhibits proliferation and induces differentiation of a variety of cells including osteoclasts, keratinocytes, HL-60 cells, vascular smooth muscle cells, and bone marrow macrophage precursors. It can activate signal transduction pathways that include hydrolysis of phosphatidylcholine and formation of diacylglycerol in chick myoblasts, and activation of Raf kinase in rat hepatic primary cells and 3T3 fibroblasts. It regulates c-myc proto-oncogene transcription, IL-8 expression, cyclooxygenase induction, and directly activates protein kinase C at physiological concentrations (EC $_{50}$ =16 nM). It up-regulates RANKL level and down-regulates OPG level.

Citations: 46

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Ordering Information

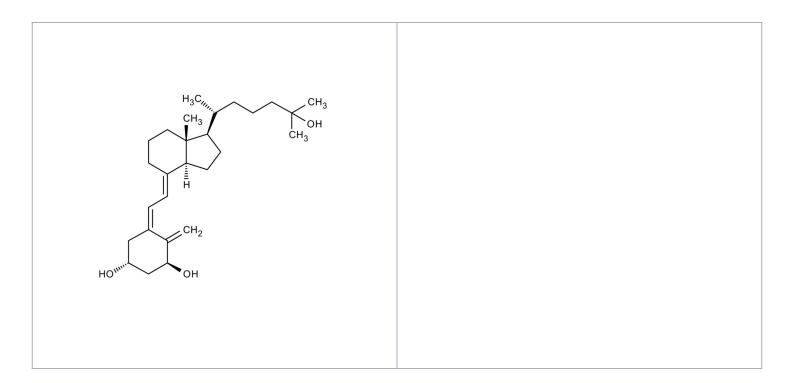
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BML-DM200-1000	1mg
BML-DM200-0050	50µg

Manuals, SDS & CofA

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- · Ultra-high purity
- Used in bone, immunology, and metabolism studies
- · Highly cited



Handling & Storage

Use/Stability As indicated on product label or CoA when stored as recommended.

Handling Protect from light. For maximum stability, store solid in the dark under an inert gas such

as argon. Store dissolved product in the dark in a tightly sealed container.

Long Term Storage -80°C

Shipping Dry Ice

Regulatory Status RUO - Research Use Only

Product Details

Alternative Name 1α,25-Dihydroxycholecalciferol, Calcitriol

Appearance White solid.

CAS 32222-06-3

Couple Target RANKL, Vitamin D receptor

Couple Type Activator, Ligand

Formula $C_{27}H_{44}O_3$

Identity Identity determined by MS and NMR.

MW 416.6

Purity ≥95% (HPLC)

Solubility Soluble in acetone, chloroform, DMSO, 100% ethanol, ethyl acetate, or tetrahydrofuran. Evaporative loss of

solvent may occur during storage and handling of the dissolved product. If necessary, exact concentration may

be re-determined spectrophotometrically.

Source



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